Degree in Civil Engineering Subject Planning



COURSE NAME

Name: MATERIALS CHARACTERIZATION

Code: 101131

Curriculum: DEGREE IN CIVIL ENGINEERING Year: 2

Name of the module to which it belongs: COMMON MODULE FOR THE CIVIL BRANCH

Subject: MATERIALS CHARACTERIZATION

Nature: OBRIGATORY Duration: FIRST SEMESTER

ECTS Credits: 6 Classroom hours: 60 Face-to-face classroom percentage: 40% Non-contact hours: 90

FACULTY DETAILS

Name: FERNÁNDEZ RODRÍGUEZ, JOSÉ MARÍA (Coordinator)

Department: INORGANIC CHEMISTRY AND CHEMICAL ENGINEERING

Area: QUÍMICA INORGÁNICA

Location of the office: Inorganic Chemistry Laboratory Office

E-Mail: um1feroj@uco.es Phone number: 618808043

Name: CANTADOR FERNÁNDEZ, DAVID

Department: INORGANIC CHEMISTRY AND CHEMICAL ENGINEERING

Area: QUÍMICA INORGÁNICA

Location of the office: Inorganic Chemistry Laboratory Office

E-Mail: p12cafed@uco.es Phone number: 680835813

SKILLS

CR1

CB2	profession of Technical Civil Engineering. Have and understand updated and cutting-edge knowledge related to the field of study of the degree of Technical Civil Engineering.
CB3	Be able to apply the knowledge acquired to their work or vocation in a professional manner. Prepare and defend arguments in the relevant knowledge area.
CU2	Know and refine the user level of ITs.
CEC2	Theoretical and practical knowledge of the chemical, physical, mechanical and technological properties of the materials most used in construction
CEC3	Ability to apply knowledge of construction materials to structural systems. Knowledge of the relationship between the structure of materials and the mechanical properties associated with their structure.

Have and understand specific knowledge of the study area of the Degree that gives skills for the evercise of the

OBJECTIVES

This course is part of the "Science and Technology of Materials" subject and of the common module of the Civil Branch. Its objective is to offer students a scientific point of view of the structure and properties of the most commonly used materials in Civil Engineering. This way, they can gain the necessary knowledge of their structure, texture, chemical composition, as well as of the reactions that can occur depending on factors such as temperature or pressure, etc.

CONTENTS:

1. Theoretical contents

Lesson 1. Structure of crystalline solids.

Lesson 2. Crystalline imperfections. Point defects.

Lesson 3. Phase diagrams. Gibbs phase rule.

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- Lesson 4. Kinetics of processes in solids.
- Lesson 5. Glasses.
- Lesson 6. Cements.
- Lesson 7. Hydration of Portland cement.
- Lesson 8. Types of Portland cements.
- Lesson 9. Pozzolanas and pozzolanic cements.
- Lesson 10. Corrosion.
- Lesson 11. Engineering Alloys.
- Lesson 12. Ceramic materials.
- Lesson 13. Polymeric materials.
- Lesson 14. Composite materials.

2. Practical contents.

- Exercise 1. Identification of crystalline phases by X-ray diffraction. X-ray program.
- Exercise 2. Identification of crystalline phases by X-ray diffraction. Example of phase identification.
- Exercise 3. Identification of crystalline phases by X-ray diffraction. Determination of mixtures.
- Exercise 4. Chloride and sulfate content in soil and water.
- Exercise 5. Ammonium content in water
- Exercise 6. Calculations in phase diagrams.
- Exercise 7. Calculations in phase diagrams. Problem solving.
- Exercise 8. Company visit.
- Exercise 9. Company visit.